Home Search Collections Journals About Contact us My IOPscience

Magnetic properties of phases possessing block type structures in the Nb_2O_{5-2x} system, with O<or=x<or=0.083

This article has been downloaded from IOPscience. Please scroll down to see the full text article. 1991 J. Phys.: Condens. Matter 3 3997 (http://iopscience.iop.org/0953-8984/3/22/011)

View the table of contents for this issue, or go to the journal homepage for more

Download details: IP Address: 171.66.16.147 The article was downloaded on 11/05/2010 at 12:09

Please note that terms and conditions apply.

Magnetic properties of phases possessing block type structures in the Nb₂O_{5-2x} system, with $0 \le x \le 0.083$

C H Rüscher and M Nygren

Institut für Mineralogie, Universität Hannover, Welfengarten 1, D-3000 Hannover, Federal Republic of Germany and Department of Inorganic Chemistry, Arrhenius Laboratory, University of Stockholm, 10691 Stockholm, Sweden

Received 15 February 1991, in final form 21 March 1991

Abstract. The magnetic susceptibility of phases with block type structures in the Nb₂O_{5-2x} system, with $0 \le x \le 0.083$, has been determined by the Faraday method in the temperature region 77–400 K. The samples consisted of crystals of composition Nb_{2x}O₇₁₁ (Nb₂O₅), Nb₅₃O₁₃₂ (NbO_{2,491}), Nb₂₅O₆₂ (NbO_{2,491}), Nb₄₃O₁₁₆ (NbO_{2,493}), Nb₂₂O₅₄ (NbO_{2,454}) and Nb₁₂O₂₉ (NbO_{2,417}). It is shown that the observed paramagnetic behaviour cannot be explained within the framework of classical Debye–Langevin behaviour or by Pauli paramagnetism. Instead, a Van Vleck type analysis of the magnetic susceptibility data gives evidence for the occurrence of bipolaron/single-polaron states at lower degrees of reduction (x < 0.05), in agreement with previous interpretations of electrical and optical properties. The quasi-metallic state for $x \ge 0.05$ reveals localized electrons, as suggested earlier.

1. Introduction

There exist several Nb₂O_{5-2x} phases, with $0 \le x \le 0.083$, that possess block-type structures (Norin and Magneli 1960, Norin 1963, 1965, 1966, Gatehouse and Wadsley 1964, Norin *et al* 1966, Gruehn and Norin 1967, 1969, Nimmo and Anderson 1972). These structures consist of blocks of $n \times m$ corner-sharing NbO₆ octahedra in the cross *a*-*b* section, indefinitely extended by corner-sharing in the direction of the unit-cell *c* axis. The blocks are interconnected by edge-sharing octahedra, yielding two sets of columns: one with its base at z = 0 and the other at z = 0.5 (c/2). The structure of H-Nb₂O₅ is outlined in figure 1. Periodical intergrowth of equally or differently sized $n \times m$ blocks determines the oxygen content of the phases. Each NbO₆ Octahedron contains one Nb atom, and additional tetrahedrally coordinated Nb atoms are found in the square tunnels formed at the corners of four interconnected blocks, at z = c/4 and 3c/4, for all phase compositions except for x = 0.083 (Nb₁₂O₂₉).

In this article we describe the magnetic properties of six different Nb₂O_{5-2x} phases: Nb₁₂O₂₉ (NbO_{2.417}), Nb₂₂O₅₄ (NbO_{2.454}), Nb₄₇O₁₁₆ (NbO_{2.468}), Nb₂₅O₆₂ (NbO_{2.480}), Nb₅₃O₁₃₂ (NbO_{2.491}) and H-Nb₂O₅. In this study, as well as in the previous one dealing with the electrical and optical properties of these phases (Rüscher *et al* 1988), we have utilized single crystals prepared by transport reaction as described by Hussain *et al* (1986). These studies have revealed that:

(i) The conductivity increases with increasing x values, i.e. degree of reduction.



3998

Figure 1. The structure of H-Nb₂O₅.

(ii) Below 250 K the temperature dependence of the conductivity changes from semiconductor to quasi-metallic behaviour with increasing x value, the compounds with $x \le 0.01$ being semiconducting. Thus Nb₅₃O₁₃₂ exhibit an apparent activation energy of 0.01 eV.

(iii) Above 250 K, the temperature dependence of the conductivity approximates $T^{-3/2}$ behaviour for all the compounds.

(iv) A broad absorption band is found in the near-infrared spectral range, with $hw_{max} \approx 0.7 \text{ eV}$, which extends asymmetrically into the visible range. The integrated intensity of this band increases with increasing degree of reduction up to $x \approx 0.05$ at T = 293 K.

The electrical and optical properties for $x \le 0.05$ could consistently be described in terms of polaron theory. The Austin and Mott (1969) description of small polarons, stating that $hw_{max} \le 4E_a$ and $E_a \simeq (1/r_p - 1/R_p)$ (e_p = polaron radius, R_p = average distance between the charge carriers), seems to be roughly obeyed for $x \le 0.01$. The surprising observation that the polaron absorption increases up to x = 0.05 although quasi-metallic behaviour sets in at $x \approx 0.02$ can be understood in terms of the polarons being highly correlated or of the carriers consisting of a mixture of bipolarons and single polarons for $0.02 \le x \le 0.05$. Finally the threshold value at $x \approx 0.05$ of the integrated intensity of the polaron absorption suggests that the quasi-metallic behaviour for compounds with x > 0.05 is due to superimposed quasi-free electrons, as previously found for WO_{3-x} with x > 0.1 (Salje and Güttler 1984).

In this article we describe the magnetic properties of the various Nb₂O_{5-2x} phases. From polaron-polaron correlation and polaron-free electron interactions one would expect strong deviation from classical Debye-Langevin behaviour and/or from Pauli paramagnetism. We are going to show that the observed paramagnetic properties can be explained within the framework for a transition from the bipolaron/single-polaron state to the quasi-metallic state with increasing x.

2. Experiment

Needle-shaped crystals of maximum sizes up to $2 \times 0.2 \times 0.2 \text{ mm}^3$, of the compositions $Nb_{12}O_{29} (NbO_{2.147})$, $Nb_{22}O_{54} (NbO_{2.454})$, $Nb_{47}O_{116} (NbO_{2.468})$, $Nb_{25}O_{62} (NbO_{2.480})$, $Nb_{53}O_{132} (NbO_{2.491})$ and $Nb_{28}O_{70} (NbO_{2.500})$ have been prepared by a transport reaction

Composition	Block-type structure	O/Nb ratio	Homogeneity range	V _N (Å ³)	N _{th}
Nb ₁₂ O ₂₉	3×4	2.417	2.471	47.71	3.5×10^{21}
Nb ₂₂ O ₅₄	$3 \times 3 + 3 \times 4 + T$	2.454	2.448-2.454	47.69	1.9×10^{21}
Nb ₄₇ O ₁₁₆	$3 \times 3 + 2(3 \times 4) + T$	2.468	2.464-2.467	47.97	1.3×10^{21}
Nb25O62	$3 \times 4 + T$	2.480	2.472-2.478	47.60	$0.8 imes 10^{21}$
Nb53O(32	$2(3 \times 4) + 3 \times 5 + T$	2.491	2.484-2.485	48.38	$0.4 imes 10^{21}$
Nb ₂₈ O ₇₀	$3 \times 4 + 3 \times 5 + T$	2.500	2.500-2.489	48.40	0

Table 1. Structural data for the Nb₂O₅₋₂ phases of concern in this study. The homogeneity ranges given are those determined by Schäfer *et al* (1969). The number of Nb⁴⁺ ions per cm³ (N_{th}) has in each case been calculated with use of the unit cell volume, V_N , of the ideal composition. *T* denotes occurrence of tetrahedrally coordinated Nb atoms in the structure.

technique using HgCl₂ as the transporting agent. A detailed description of the preparatory route used and of the determination of the degree of reduction, i.e. x in Nb₂O_{5-2x}, is given by Hussain *et al* (1986). It has previously been shown that these phases are stoichiometric or have very limited homogeneity ranges with respect to the Nb/O ratio (Schäfer *et al* 1969). These ranges, together with some structural data and the calculated numbers of Nb⁴⁺ ions per cm³ for the various phases are given in table 1.

The magnetic susceptibility of about 20–30 mg of as-grown crystals of each composition was determined by the Faraday method. The field strength was $H = 0.7 \times 10^7$ A m⁻¹, and the resolution of the weight measurements both with and without magnetic field was better than 5 µg. The set up was calibrated with CoHg(SCN)₄ and the measurements were carried out in the temperature region 77–400 K. Each measured point represents the average of three switch on-switch off cycles of the magnetic field at a fixed temperture, and data were recorded for increasing T. Successive runs of the same samples showed the reproducibility of the measurements to be better than 1%. The crystals had a good statistical distribution of orientation within the sample holder, i.e. we were not able to measure any anisotropic susceptibility by different modes of packing.

3. Results and discussion

The molar magnetic susceptibilities of the compounds under investigation are given in figure 1. All compounds exhibited paramagnetic behaviour. The susceptibilities of NbO_{2.500} and NbO_{2.491} were found to be temperature independent. For the other compounds, both the susceptibility at a given temperature and the temperature dependence of the susceptibility increase with increasing degree of reduction, i.e. with increasing concentration of Nb⁴⁺ ions present. The increase of χ_m at 100, 200 and 300 K with the degree of reduction (x in Nb₂O_{5-2x}) is illustrated in figure 2. According to figure 2, it seems that the susceptibility saturates for the compounds with $x \ge 0.05$. The compositional variation of the susceptibility is so large that any influences not related to x, such as impurity effects, diamagnetic contributions of the Nb and O ions and small deviations from the ideal stoichiometry (see table 1), can be neglected.

4000

C H Rüscher and M Nygren



Figure 2. The molar magnetic susceptibility, χ_m in emu mol⁻¹, plotted as a function of the temperature, T, for Nb₂₄O₇₀ (NbO_{2.500}), Nb₅₃O₁₃₂ (NbO_{2.491}), Nb₂₅O₆₂ (NbO_{2.480}), Nb₄₂O₁₁₆ (NbO_{2.468}), Nb₂₂O₅₄ (NbO_{2.454}) and Nb₁₂O₂₉ (NbO_{2.417}). The inset shows the temperature dependence of $1/\chi_m$ for NbO_{2.417}, NbO_{2.454} and NbO_{2.468}.

As the compounds with $x \ge 0.02$ exhibited quasi-metallic behaviour one would expect them to exhibit Pauli paramagnetism, which can be estimated from the formula:

$$X_{\rm p} = g(\mathcal{E}_{\rm F})\mu_{\rm B}^2 N_{\rm a}/N \tag{1}$$

where $g(E_F)$ = the density of states at the Fermi energy, μ_B = the Bohr magneton, N_a = 6.023 × 10²³ (Avogadro number) and N = the total carrier concentration. We have calculated X_p values for the various compounds and plotted them in figure 2, assuming the concentration of free electrons (N) to be equal to the number of Nb⁴⁺ ions present in each compound (N_{th} in table 1). A further assumption is an isotropic Fermi distribution, implying that:

$$g(E_{\rm F}) = m^* / h \pi^2 (3N\pi)^{1/3}.$$
 (2)

In the calculations, the effective mass, m^* , has been set equal to one. As seen in figure 3, our measured data are clearly at variance with the calculated ones, implying that we should eliminate Pauli paramagnetism from further considertion.

The temperature dependence of the susceptibility of all compounds except $Nb_{12}O_{29}$ ($NbO_{2,417}$) and $Nb_{22}O_{54}$ ($NbO_{2,454}$), could be fitted to the equation:

$$X_{\rm m} = C/T + B \tag{3}$$

with $C = N_a \mu_B^2 \langle (\mu_p)^2 \rangle / 3k$ and B equal to the temperature-independent paramagnetic contribution as seen in figure 3 where f(T) defined as

$$f(T) = \chi_{\rm m} T 3k / (N_{\rm a} \mu_{\rm B}^2) \tag{4}$$

is plotted against the temperature.

Following van Vleck (1965), B in (3) denotes the temperature-independent paramagnetism resulting from excitations from states with energetical separations exceeding kT, $\delta E \ge kT$, i.e. high frequency elements. B can be determined from the slopes of the f(T) versus T full curves in figure 3. In our case the high-frequency excitations could

Composition	Degree of reduction $x \text{ in } Nb_2O_{5-2x}$	$\mu_{ m catc}$	$\mu_{\rm obs}$
Nb ₁₂ O ₂₉	0.083	0.693	≅0.2
Nb ₂₂ O ₅₄	0.046	0.525	≅0.1
Nb ₄₂ O ₁₁₆	0.032	0.438	0.045
Nb25O62	0.020	0.346	0.010
Nb ₅₃ O ₁₃₂	0.009	0.232	< 0.005
Nb28070	0.000	0.000	<0.005



Figure 3. The molar magnetic susceptibility (χ_m) at 100, 200 and 300 K plotted as a function of the degree of reduction x in the system Nb₂O_{5-2x}. The Pauli susceptibilities (χ_p) calculated according to equations 1 and 2 (see text) are also given. The curves are given as guides for the eyes.

arise from phonon-assisted hopping of polarons. In the van Vleck picture, the f(T = 0 K) quantity results from low-frequency elements (normal states) with $\delta E \leq kT$ of permanent moments. The permanent moments (μ_p) can be calculated according to:

$$\mu_{\text{calc}} = g[J(J+1)2x]^{1/2} \qquad g = 2 \text{ and } J = \frac{1}{2}.$$
(5)

Assuming that each Nb⁴⁺ ion contributes one moment, and neglecting spin-orbit coupling effects one obtains the μ_{calc} values given in table 2. The μ_{obs} values give in the same table are extrapolations to T = 0 of the f(T) versus T curves in figure 3. Consistently $\mu_{obs} \ll \mu_{calc}$, which can be understood within the framework of the bipolaron picture. The increasing number of effective magnetons, μ_p , with increasing temperature means breaking of bipolaron states. The observation that $\mu_{obs} \neq 0$ at 0 K is then due to saturation of the bipolaron-order parameter (Rüscher *et al* 1990).





The susceptibility of Nb₁₂O₂₉ follows the Curie-Weiss behaviour

$$\chi_{\rm m} = C/(T - \theta) \tag{6}$$

Nb₁₂O₂₉ (NbO_{2,417}).

with $\theta \approx 47$ K (see inset in figure 1). From the slope of $(\chi_m)^{-1} = f(T)$ curve we calculate C and deduce to $\langle (\mu_p)^2 \rangle^{1/2}$ value of the order 0.5. The corresponding x value calculated according to (5) corresponds to $x \approx 0.042$, implying that for this phase composition, about 50% of all possible moment contribute to the susceptibility. Corresponding data for Nb₂₂O₅₄ (with $\theta \approx 120$ K) show that all Nb⁴⁺ ions introduce freely rotating moments. From structural considerations it seems plausible that one cannot introduce more than one Nb⁴⁺ ion possessing a freely rotating moment per block of 3×4 NbO₆ octahedra. Introducing more moments thus implies that the structural capability of hosting freely rotating moments is exceeded. This is in agreement with the observation that the integrated intensity of the optical absorption increases with increasing degree of reduction up to $x \approx 0.05$ (Rüscher *et al* 1988). The optical and magnetic properties of Nb₂₂O₅₄ (NbO_{2.454}) can thus be interpreted in terms of the concentration of polarons saturating around x = 0.05. When the degree of reduction, x, exceeds 0.05 quasi-free electrons start to appear. Accordingly in Nb₁₂O₂₉ (NbO_{2.417}) the principal charge carriers ought to consist of a mixture of polarons and quasi-free electrons.

4. Concluding remarks

The considerable reduction of the effective magnetic moments in the compounds described above can be understood in terms of these compounds possessing a mixture of bipolaron-single polaron states. Pair breaking occurs with increasing degree of reduction resulting in increasing $\langle (\mu_p)^2 \rangle$ values as seen in figure 3. It has previously been shown that the extension of the wave function of the charge carriers is closely related to the size of the structural block unit (Ruscher *et al* 1988). This implies that when chemical reduction introduces more than one electron per block of 3×4 NbO₆ octahedra, as in Nb₁₂O₂₉, one would expect the appearance of quasi-free electrons. The magnetic as well as the electrical properties of Nb₁₂O₂₉ thus seem to be understandable in terms of superimposition of localized electrons (polarons) and free carriers.

Acknowledgments

We are grateful to Professors R Gruehn and A Hussain for providing us with the samples. This work has been supported in part by SFB 173, BMFT 13N57387 and the Swedish Natural Science Foundation.

References

Austin I G and Mott N F 1969 Adv. Phys. 18 41

Gatehouse B M and Wadsley A D 1964 Acta. Crystallogr. 17 1545

Gruehn R and Norin R 1967 Z. Anorg. Allg. Chemie 355 176

Gruehn R and Norin R 1969 Z. Anorg. Allg. Chemie 367 209

Hussain A, Reitz B and Gruehn R 1986 Z. Anorg. Allg. Chemie 535 186

Nimmo K and Anderson J S 1972 J. C. S. Dalton 2328

Norin R 1963 Acta. Chem. Scand. 17 1391

Norin R 1965 Naturwissenschaften 52 300

Norin R 1966 Acta. Chem. Scand. 20 871

Norin R, Carlson M and Eleqvist B 1966 Acta Chem. Scand. 20 2892

Norin R and Magneli A 1960 Naturwissenschaften 47 354

Rüscher CH, Rüffer N, Seiwert FJ and Gruehn R 1990 Physica C 166 158

Rüscher CH, Salje E and Hussain A 1988 J. Phys. C: Solid State Phys. 21 3737

Salje E and Güttler B 1984 Phil. Mag. B 50 607

Schäfer H, Bergner D and Gruehn R 1969 Z. Anorg. Allg. Chemie 365 31

Van Vleck J H 1965 The Theory of Electric and Magnetic Susceptibilites 1st edn. (Oxford: Oxford University Press) p 181